

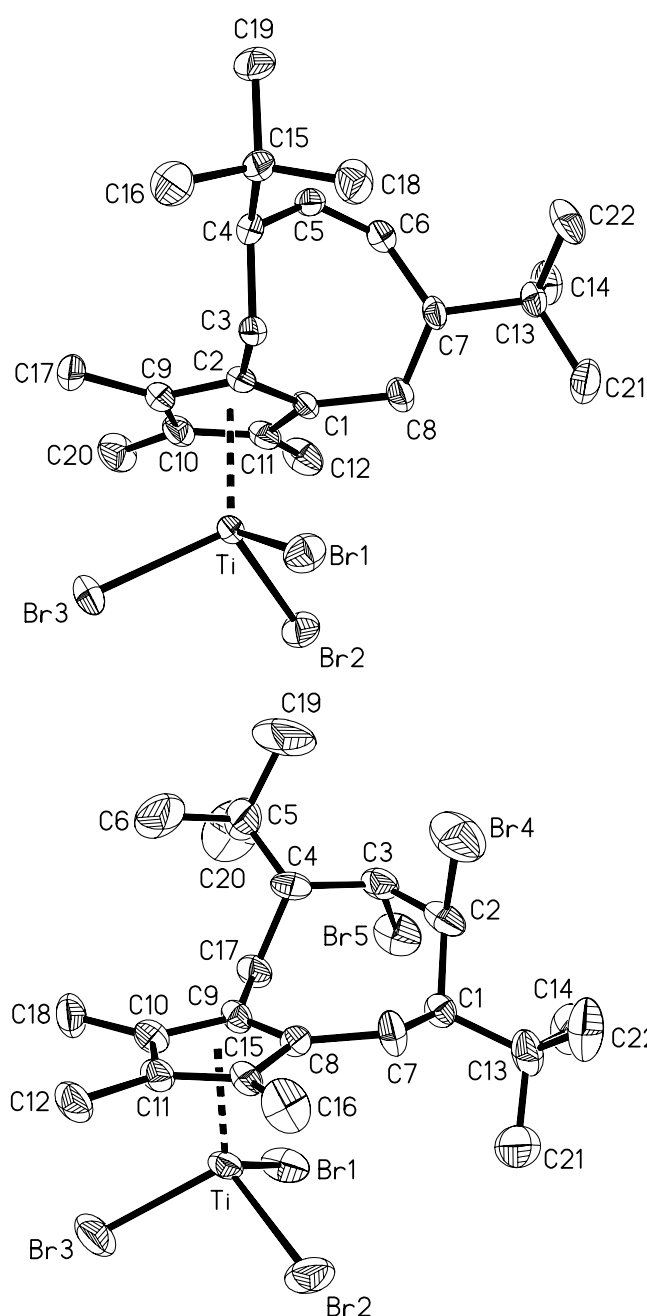
Crystal structures of tribromo(η^5 -3,6-di-*tert*-butyl-9,10,11-trimethylbicyclo(6.3.0)undeca-4-en-8,10-dienyl)titanium(IV), $\text{Ti}(\text{C}_{22}\text{H}_{35})\text{Br}_3$, and tribromo(η^5 -4,5-dibromo-3,6-di-*tert*-butyl-9,10,11-trimethylbicyclo(6.3.0)undeca-8,10-dienyl)titanium(IV), $\text{Ti}(\text{C}_{22}\text{H}_{35}\text{Br}_2)\text{Br}_3$

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Abstract

$\text{C}_{22}\text{H}_{35}\text{Br}_3\text{Ti}$, triclinic, $P\bar{1}$ (no. 2), $a = 9.621(2)$ Å, $b = 11.796(2)$ Å, $c = 12.232(2)$ Å, $\alpha = 102.23(3)^\circ$, $\beta = 97.71(3)^\circ$, $\gamma = 112.32(3)^\circ$, $V = 1219.2$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.058$, $wR_{\text{obs}}(F^2) = 0.134$, $T = 293$ K.

$\text{C}_{22}\text{H}_{35}\text{Br}_5\text{Ti}$, monoclinic, $P12_1/n1$ (no. 14), $a = 7.474(1)$ Å, $b = 18.458(4)$ Å, $c = 20.171(4)$ Å, $\beta = 100.28(3)^\circ$, $V = 2738.0$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.054$, $wR_{\text{obs}}(F^2) = 0.119$, $T = 293$ K.

Source of material

Dibromo-pentamethylcyclopentadienyl-trimethylcyclopentadienyl-cyclooctene-Ti(IV) [1] (**1**; 125 mg, 0.195 mmol) was dissolved in 20 ml of *n*-hexane under argon and treated with an excess (0.3 ml) of bromine. Standing at room temperature for 20 h and concentration of the orange-red solution in vacuum to 2 ml gave a mixture of red crystals of complexes Cp^*TiBr_3 , the title compounds (**2** and **3**) in a molar ratio of 35:35:30 as identified by NMR. Fractional recrystallization from *n*-hexane gave small amounts of individual complex **2** (20 mg) and complex **3** (30 mg) which were separated from the mother liquor, washed with cold *n*-hexane and dried in vacuum.

Discussion

For group 4 decamethyltitanocenes and -zirconocenes manifold C—H bond activations of methyl groups of pentamethylcyclopentadienyl ligands with coupling reactions with butadiynes $RC\equiv C-C\equiv CR$ have been verified for $M = \text{Ti}$, $R = \text{Me}$, Ph , $t\text{-Bu}$ and $M = \text{Zr}$, $R = t\text{-Bu}$ [2-4]. This is a facile method for functionalizing metallocenes to obtain chiral and non-chiral complexes [1,5]. One example for this is a formal Ti(II) complex with one pentamethylcyclopentadienyl and one annellated trimethylcyclopentadienyl-cyclooctyne ligand [1-5]. Hydrogenation of this complex yields a Ti(II) complex with one pentamethylcyclopentadienyl and one annellated trimethylcyclopentadienyl-cyclooctene ligand, which is easily oxidized by bromine to a Ti(IV) dibromide (**1**) with one pentamethylcyclopentadienyl and one annellated trimethylcyclopentadienylcyclooctene ligand [1].

In the crystal structures of the title compounds, the η^5 -coordinated Cp ligand and three bromine ligands form a "piano-stool"-like coordination. As in the case of the starting complex **1** and their chloride congener the cyclooctene unit is oriented away from the metal. There is no hint for any interaction of the double bond with the metal in **2**. This was expected for the electronically saturated Ti(IV). The bromine substituents in **3** are in an *anti* conformation to each other.

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1. Tribromo(η^5 -3,6-di-*tert*-butyl-9,10,11-trimethyl-bicyclo(6.3.0)undeca-4-en-8,10-dienyl)titanium(IV), Ti(C₂₂H₃₅)Br₃

Table 1. Data collection and handling.

Crystal:	red prism, size 0.3 × 0.3 × 0.4 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	52.72 cm ⁻¹
Diffractometer, scan mode:	Stoe IPDS I, φ
$2\theta_{\max}$:	48.5°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	3626, 3626
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2206
$N(\text{param})_{\text{refined}}$:	235
Programs:	SHELXS-86 [6], SHELXL-93 [7]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	U_{iso}
H(3A)	2i	0.2923	0.3888	0.7397	0.046
H(3B)	2i	0.2379	0.3911	0.8552	0.046
H(4)	2i	-0.0112	0.2580	0.7585	0.047
H(5)	2i	-0.0987	0.1513	0.5717	0.045
H(6)	2i	0.0313	0.1081	0.4587	0.048
H(7)	2i	0.2958	0.3285	0.5744	0.046
H(8A)	2i	0.3396	0.1024	0.5434	0.065
H(8B)	2i	0.4581	0.2398	0.6173	0.065
H(12A)	2i	0.3103	-0.1296	0.7125	0.105
H(12B)	2i	0.4270	-0.0362	0.6594	0.105
H(12C)	2i	0.2513	-0.1083	0.5961	0.105
H(14A)	2i	0.2816	0.0575	0.3559	0.098
H(14B)	2i	0.2571	0.1113	0.2518	0.098
H(14C)	2i	0.1234	0.0628	0.3143	0.098
H(16A)	2i	0.1377	0.5266	0.8749	0.097
H(16B)	2i	-0.0364	0.4422	0.8673	0.097
H(16C)	2i	0.0091	0.5670	0.8280	0.097
H(17A)	2i	0.1250	0.2679	0.9605	0.092

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	U_{iso}
H(17B)	2i	0.1900	0.1958	1.0324	0.092
H(17C)	2i	0.0281	0.1221	0.9449	0.092
H(18A)	2i	0.2282	0.5374	0.6898	0.089
H(18B)	2i	0.0965	0.5778	0.6504	0.089
H(18C)	2i	0.1087	0.4604	0.5711	0.089
H(19A)	2i	-0.2171	0.3152	0.6774	0.085
H(19B)	2i	-0.1612	0.3256	0.5638	0.085
H(19C)	2i	-0.1734	0.4430	0.6432	0.085
H(20A)	2i	0.1653	-0.0336	0.9644	0.105
H(20B)	2i	0.2975	-0.0656	0.9241	0.105
H(20C)	2i	0.1279	-0.1376	0.8477	0.105
H(21A)	2i	0.5198	0.2613	0.4532	0.101
H(21B)	2i	0.5083	0.3929	0.4738	0.101
H(21C)	2i	0.4922	0.3138	0.3487	0.101
H(22A)	2i	0.2566	0.3991	0.4107	0.101
H(22B)	2i	0.1083	0.2725	0.3481	0.101
H(22C)	2i	0.2420	0.3210	0.2856	0.101

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
C(1)	2i	0.3020(9)	0.1513(8)	0.7023(6)	0.043(4)	0.055(5)	0.025(4)	0.029(4)	0.007(4)	0.014(4)
C(2)	2i	0.2471(8)	0.2214(7)	0.7824(6)	0.033(4)	0.047(5)	0.034(4)	0.017(4)	0.004(4)	0.016(4)
C(3)	2i	0.2197(8)	0.3386(8)	0.7775(6)	0.038(4)	0.055(5)	0.029(4)	0.025(4)	0.010(4)	0.012(4)
C(4)	2i	0.0549(9)	0.3017(8)	0.7123(7)	0.036(4)	0.051(5)	0.038(4)	0.020(4)	0.016(4)	0.018(4)
C(5)	2i	0.0070(8)	0.2028(7)	0.5964(7)	0.032(4)	0.040(5)	0.044(5)	0.016(4)	0.010(4)	0.019(4)
C(6)	2i	0.0887(9)	0.1765(8)	0.5240(7)	0.046(5)	0.038(5)	0.034(4)	0.015(4)	0.007(4)	0.012(4)
C(7)	2i	0.2585(9)	0.2388(8)	0.5298(6)	0.053(5)	0.049(5)	0.025(4)	0.029(4)	0.017(4)	0.015(4)
C(8)	2i	0.350(1)	0.1808(9)	0.5957(7)	0.064(6)	0.085(7)	0.034(5)	0.045(6)	0.020(4)	0.025(5)
C(9)	2i	0.2064(9)	0.1544(8)	0.8662(7)	0.042(5)	0.062(6)	0.039(5)	0.024(5)	0.009(4)	0.024(5)
C(10)	2i	0.236(1)	0.0442(8)	0.8367(7)	0.050(5)	0.048(6)	0.043(5)	0.016(5)	0.004(4)	0.022(5)
C(11)	2i	0.2943(9)	0.0420(8)	0.7360(7)	0.050(5)	0.051(6)	0.039(5)	0.029(5)	0.002(4)	0.013(5)
C(12)	2i	0.323(1)	-0.0682(9)	0.6699(8)	0.094(8)	0.066(7)	0.053(6)	0.047(7)	0.003(6)	0.006(6)
C(13)	2i	0.294(1)	0.2392(8)	0.4074(6)	0.055(5)	0.050(6)	0.028(4)	0.024(5)	0.016(4)	0.012(4)
C(14)	2i	0.233(1)	0.1053(9)	0.3247(8)	0.086(7)	0.068(7)	0.039(5)	0.029(6)	0.024(5)	0.009(5)
C(15)	2i	0.0168(9)	0.4179(8)	0.7078(7)	0.043(4)	0.055(6)	0.044(5)	0.031(4)	0.017(4)	0.020(4)
C(16)	2i	0.033(1)	0.496(1)	0.8312(8)	0.076(7)	0.070(7)	0.060(6)	0.046(6)	0.021(6)	0.010(6)
C(17)	2i	0.131(1)	0.188(1)	0.9594(7)	0.059(6)	0.100(8)	0.051(6)	0.045(6)	0.030(5)	0.041(6)
C(18)	2i	0.122(1)	0.5066(9)	0.6493(8)	0.072(6)	0.059(6)	0.067(6)	0.035(6)	0.029(5)	0.035(6)
C(19)	2i	-0.149(1)	0.3712(9)	0.6420(8)	0.055(5)	0.066(6)	0.072(6)	0.043(5)	0.022(5)	0.029(6)
C(20)	2i	0.204(1)	-0.057(1)	0.8989(9)	0.081(7)	0.064(7)	0.072(7)	0.028(6)	0.014(6)	0.040(6)
C(21)	2i	0.470(1)	0.308(1)	0.4222(8)	0.066(6)	0.076(7)	0.055(6)	0.017(6)	0.031(5)	0.023(6)
C(22)	2i	0.218(1)	0.315(1)	0.3585(8)	0.100(8)	0.070(7)	0.052(6)	0.046(7)	0.021(6)	0.037(6)
Ti	2i	0.4748(2)	0.2251(1)	0.8833(1)	0.0383(8)	0.051(1)	0.0299(7)	0.0217(7)	0.0079(6)	0.0145(7)
Br(1)	2i	0.6019(1)	0.43937(9)	0.87439(9)	0.0515(5)	0.0515(6)	0.0720(7)	0.0178(5)	0.0151(5)	0.0202(5)
Br(2)	2i	0.6596(1)	0.1422(1)	0.84112(9)	0.0587(6)	0.0765(7)	0.0735(7)	0.0439(6)	0.0231(5)	0.0277(6)
Br(3)	2i	0.4996(1)	0.2475(1)	1.08485(8)	0.0702(7)	0.1041(9)	0.0324(5)	0.0286(7)	0.0103(5)	0.0204(6)

2. Tribromo(η^5 -4,5-dibromo-3,6-di-*tert*-butyl-9,10,11-trimethylbicyclo(6.3.0)undeca-8,10-dienyl)titanium(IV), Ti(C₂₂H₃₅Br₂)Br₃

Table 4. Data collection and handling.

Crystal:	red prism, size 0.1 × 0.1 × 0.3 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	76.16 cm ⁻¹
Diffractometer, scan mode:	Stoe IPDS I, φ
$2\theta_{\max}$:	42.18°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	5432, 2818
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 1469
$N(\text{param})_{\text{refined}}$:	253
Programs:	SHELXS-86 [6], SHELXL-93 [7]

Table 5. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	4e	-0.1495	0.2286	0.2983	0.056
H(2)	4e	0.0335	0.2631	0.4080	0.083
H(3)	4e	0.1812	0.1546	0.4296	0.073
H(4)	4e	0.2414	0.1314	0.3181	0.070
H(6A)	4e	0.3985	0.0312	0.2996	0.196
H(6B)	4e	0.2254	-0.0179	0.2822	0.196
H(6C)	4e	0.3898	-0.0428	0.3370	0.196
H(7A)	4e	-0.0431	0.3053	0.2132	0.074
H(7B)	4e	0.1494	0.2887	0.2542	0.074
H(12A)	4e	0.1497	0.0729	0.0322	0.096
H(12B)	4e	0.2800	0.1401	0.0420	0.096
H(12C)	4e	0.0796	0.1486	0.0035	0.096
H(14A)	4e	-0.3050	0.2939	0.3828	0.144
H(14B)	4e	-0.2900	0.3785	0.3885	0.144
H(14C)	4e	-0.1262	0.3298	0.4213	0.144
H(16A)	4e	0.1261	0.3202	0.1328	0.118

Table 5. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(16B)	4e	0.0758	0.2876	0.0601	0.118
H(16C)	4e	0.2724	0.2774	0.1017	0.118
H(17A)	4e	-0.1240	0.1091	0.2882	0.068
H(17B)	4e	-0.0220	0.0398	0.2695	0.068
H(18A)	4e	0.0849	-0.0050	0.1016	0.115
H(18B)	4e	-0.0506	-0.0148	0.1519	0.115
H(18C)	4e	0.1595	-0.0125	0.1790	0.115
H(19A)	4e	0.4690	0.0918	0.4146	0.237
H(19B)	4e	0.4565	0.0153	0.4472	0.237
H(19C)	4e	0.3373	0.0794	0.4659	0.237
H(20A)	4e	0.0627	0.0073	0.4328	0.231
H(20B)	4e	0.1876	-0.0566	0.4175	0.231
H(20C)	4e	0.0232	-0.0317	0.3627	0.231
H(21A)	4e	-0.3898	0.3092	0.2589	0.171
H(21B)	4e	-0.2650	0.3552	0.2202	0.171
H(21C)	4e	-0.3715	0.3930	0.2710	0.171

Table 6. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
C(1)	4e	-0.053(2)	0.2651(7)	0.3069(6)	0.05(1)	0.05(1)	0.041(8)	-0.004(8)	0.013(7)	-0.004(7)
C(2)	4e	0.077(2)	0.2394(9)	0.3705(7)	0.05(1)	0.10(1)	0.05(1)	-0.017(9)	0.000(8)	-0.016(9)
C(3)	4e	0.089(2)	0.1588(8)	0.3888(6)	0.07(1)	0.08(1)	0.035(9)	0.002(9)	0.002(7)	0.002(8)
C(4)	4e	0.146(2)	0.1065(7)	0.3370(6)	0.06(1)	0.07(1)	0.033(9)	0.003(8)	-0.006(8)	0.019(8)
C(5)	4e	0.236(3)	0.037(1)	0.3720(8)	0.13(2)	0.11(2)	0.04(1)	0.04(1)	0.02(1)	0.00(1)
C(6)	4e	0.320(3)	-0.002(1)	0.3176(9)	0.17(2)	0.13(2)	0.09(1)	0.08(2)	0.00(1)	0.03(1)
C(7)	4e	0.027(2)	0.2695(7)	0.2419(6)	0.09(1)	0.05(1)	0.049(9)	-0.008(9)	0.023(8)	-0.015(8)
C(8)	4e	0.039(2)	0.2034(7)	0.1995(6)	0.04(1)	0.04(1)	0.035(8)	0.002(7)	0.011(6)	0.004(7)
C(9)	4e	0.022(2)	0.1293(8)	0.2111(6)	0.05(1)	0.06(1)	0.032(8)	-0.004(8)	0.013(6)	-0.006(8)
C(10)	4e	0.053(2)	0.0886(8)	0.1536(7)	0.06(1)	0.06(1)	0.034(9)	-0.001(8)	0.007(7)	0.001(9)
C(11)	4e	0.094(2)	0.1420(8)	0.1067(6)	0.05(1)	0.07(1)	0.031(8)	0.006(8)	0.010(7)	-0.003(9)
C(12)	4e	0.157(2)	0.1242(8)	0.0398(6)	0.07(1)	0.09(1)	0.034(8)	0.002(9)	0.013(7)	-0.014(8)
C(13)	4e	-0.149(3)	0.3371(9)	0.3186(7)	0.11(2)	0.06(1)	0.05(1)	-0.00(1)	0.03(1)	-0.015(9)
C(14)	4e	-0.225(3)	0.3346(9)	0.3838(7)	0.13(2)	0.11(2)	0.07(1)	0.02(1)	0.05(1)	-0.01(1)
C(15)	4e	0.087(2)	0.2106(8)	0.1335(6)	0.04(1)	0.05(1)	0.039(8)	-0.010(8)	0.005(7)	-0.010(8)
C(16)	4e	0.146(2)	0.2803(8)	0.1044(7)	0.09(1)	0.08(1)	0.07(1)	-0.03(1)	0.03(1)	0.003(9)
C(17)	4e	-0.011(2)	0.0916(7)	0.2770(6)	0.07(1)	0.06(1)	0.032(8)	0.003(9)	0.006(7)	0.005(7)
C(18)	4e	0.063(2)	0.0064(7)	0.1458(7)	0.12(2)	0.05(1)	0.06(1)	0.02(1)	0.013(9)	-0.012(8)
C(19)	4e	0.389(3)	0.058(1)	0.4304(9)	0.17(3)	0.19(2)	0.09(2)	0.03(2)	-0.06(2)	0.03(2)
C(20)	4e	0.117(3)	-0.016(1)	0.399(1)	0.21(3)	0.13(2)	0.14(2)	0.02(2)	0.09(2)	0.08(2)
C(21)	4e	-0.308(3)	0.350(1)	0.2620(8)	0.12(2)	0.14(2)	0.08(1)	0.05(1)	0.00(1)	-0.02(1)
C(22)	4e	-0.025(3)	0.4009(9)	0.3223(9)	0.16(2)	0.06(1)	0.11(2)	-0.01(1)	0.06(1)	-0.01(1)
Ti	4e	-0.2094(3)	0.1593(1)	0.1190(1)	0.044(2)	0.069(2)	0.028(1)	0.004(1)	0.002(1)	0.000(1)
Br(1)	4e	-0.4247(2)	0.1385(1)	0.19187(7)	0.047(1)	0.134(2)	0.055(1)	-0.007(1)	0.0135(8)	0.010(1)
Br(2)	4e	-0.3121(3)	0.2726(1)	0.06958(8)	0.097(2)	0.104(2)	0.058(1)	0.037(1)	0.0119(9)	0.029(1)
Br(3)	4e	-0.3079(3)	0.0753(1)	0.02995(8)	0.074(2)	0.131(2)	0.070(1)	-0.008(1)	-0.0101(9)	-0.045(1)
Br(4)	4e	0.3282(3)	0.2721(1)	0.3769(1)	0.083(2)	0.151(2)	0.118(2)	-0.037(1)	0.009(1)	-0.026(2)
Br(5)	4e	-0.1478(3)	0.1338(1)	0.41593(8)	0.095(2)	0.135(2)	0.061(1)	-0.012(1)	0.033(1)	0.010(1)

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